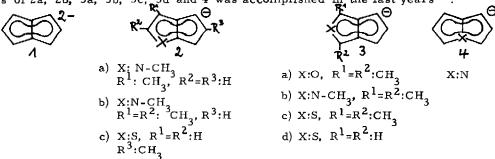
THE 5-METHYL-1-THIAPENTALENYL ANION 1)

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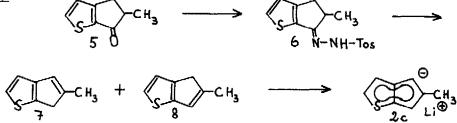
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Anionic nonbenzenoid aromatic heterocycles with 10π -electrons are of considerable interest. They are isoelectronic with the cyclooctatetraene dianion and the pentalene dianion 1^{2} . The synthesis of 2a, 2b, 3a, 3b, 3c, 3d and 4 was accomplished in the last years 3^{3} .

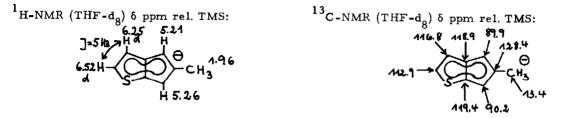


In this communication we want to report on the first synthesis of 2c. Synthesis



Ketone 5⁴⁾ reacts with tosylhydrazine in ethanolic HCl to give tosylhydrazone 6 in 93% yield. The etheral suspension of 6 was treated with 2, 5 equivalents of butyllithium (in hexane) and the mixture refluxed for 1 hr. Hydrolysis and workup gave the two olefins 7 and 8 (41%). The olefins are sensitive to air and acids. Addition of 1.17 equivalent of butyllithium (in hexane) to a solution of 7 and 8 in THF at -20° , stirring for 10 min. at this temperature, followed by warming to 0° C gave the 5-methyl-1-thiapentalenyl anion 2c. At -78° the anion 2c crystallizes in pale yellow crystals. 2c is stable for weeks at -78° C.

NMR-spectra of 2c



Reaction of 2c with methyl iodide gave the two olefins 9a and 10a (57%; mass spectrum: M^+ -peak at m/e = 150 has I_{rel} 72%). Hydrolysis of 2c with D_2O yielded the olefins 9b and 10b.



<u>Discussion</u>: Due to the ring current in 2c the nmr-signals for H_4 and $H_6(5.21 \text{ resp. } 5.26 \text{ ppm})$ appear at relatively low field. The nmr-spectrum of the isoelectronic pentalene dianion shows the signals for the protons H_1 , H_3 , H_4 and H_6 at δ 4.98. The signals of the thiophene protons of 2c are shifted upfield 0.6-0.7 ppm relative to the thiophene protons of the olefin 8. This indicates that negative charge is delocalized to some extent into the thiophene ring in 2c. This charge delocalization is also documented by the ¹³C-nmr-spectrum. The signals for C_2 and C_3 of the anion 2c are shifted upfield 10-12 ppm relative to thiophene (C_a 125.4; C_B 127.2 ppm)⁵. The ¹³C-nmr-spectrum shows furthermore that the greatest charge density resides at carbon atoms 4 and 6. The ¹³C-chemical shifts of C_4 and C_6 in 2c are of the same order as of C_1 and C_3 in the indenyl anion (91.8 ppm).

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